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Nonperturbative theory for the dispersion self-energy of atoms

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We go beyond the approximate series expansions used in the dispersion theory of finite-size atoms. We demonstrate that a correct, and nonperturbative, theory dramatically alters the dispersion self-energies of atoms. The nonperturbed theory gives as much as 100% corrections compared to the traditional series-expanded theory for the smaller noble gas atoms.

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Until recently most work, using either the quantum electrodynamic (QED) or the semiclassical formalism of dispersion energies, have relied upon a point dipolar description. Previous work that did indeed incorporate the finite atomic size was based on series expansions, and suggested that the van der Waals interaction contributed to a short-range attractive binding energy [1,2]. The aim of a recent work [3] was to demonstrate how keeping the full nonperturbative theory and taking the finite atomic size into account strongly alters the nonretarded van der Waals force and resonance interaction energy at contact distances. Including finite atomic size effects in a nonperturbative theory opens up for the possibility of having van der Waals repulsion when two atoms come very close. Series expansion is a valid approach if the coupling is very weak.

In the present Brief Report we present a derivation of the nonperturbative retarded self-energy of atoms and ions in vacuum. Within our theory, taken without series expansions, we find substantial corrections to the self-energy of an atom. A useful analytical asymptote is derived and we present some illuminating numerical results considering the finite-size effects for different atoms and ions.

In the contributions by Mahanty and Ninham [1,2] they demonstrated that the secular equation that gives the perturbed eigenmodes of the electromagnetic field due to the presence of a polarizable particle with finite size is

$$\mathbf{I} + 4\pi \tilde{G}(\vec{r}, \vec{r}', \xi) = 0,$$

where

$$\tilde{G}(\vec{r}, \vec{r}', \xi) = \left\{ \frac{\xi^2}{c^2} \mathbf{I} + \nabla_r \nabla_{r'} \right\} \tilde{G}(\vec{r} - \vec{r}', \xi) \times a(\vec{r}'' - \vec{r}', \xi) \tilde{r}'' d^3r'',$$

and the free space Green function has the form [2]

$$\tilde{G}(\vec{r} - \vec{r}', \xi) = \frac{I}{(2\pi^3)^3} \int d^3k \exp\left[ ik(\vec{r} - \vec{r}') \right] \left( \frac{\xi^2}{c^2} - k^2 \right),$$

where $\mathbf{I}$ is the unit tensor.

The dispersion energy of a finite-size isotropic molecule with a Gaussian spread was derived by Mahanty and Ninham [1,2] in a series-expanded theory. Here, we present a nonperturbative theory that can be used for atoms and ions in vacuum, and we show that there are often large corrections to the results from the approximate series-expanded theory. The dispersion self-energy is [1,2]

$$E_s = \hbar \int_0^\infty \frac{d\xi}{2\pi} \ln[I + 4\pi \tilde{G}(\vec{r}, \vec{r}, \xi)].$$

As was pointed out by Mahanty and Ninham the finite spread of the polarization make the Green function, $\tilde{G}(\vec{r}, \vec{r}, \xi)$, converge (here $\vec{r}$ is the position of the polarization). The polarization cloud of real atoms has a finite spread and in vacuum, and we consider as an interesting case a spatial distribution of the atom following an isotropic Gaussian function. This gives a Green’s tensor where the diagonal elements are equal (i.e., $j = x, y,$ and $z$ components are equal) [2,4]. The choice for the polarizability tensor is

$$\tilde{a}(\vec{r}, \xi) = I(\pi)^{3/2} a^{-3} e^{-r^2/a^2} a(\xi),$$

where $a$ is the Gaussian radius. The fully retarded expression can be shown to be

$$G_{j\alpha}(i\xi) = \frac{a(i\xi)}{3(2\pi)^3} J(i\xi),$$

where

$$I(i\xi) = \frac{\pi^{3/2}}{(a/2)^3} \left( 1 + \frac{\xi^2 a^2}{c^2} \right) - 4\pi^2 \frac{\xi^3}{c^2} e^{(a/2c)^2} \left[ 1 - \text{erf} \left( \frac{\xi a}{2c} \right) \right].$$

In the nonretarded limit this is reduced to

$$G_{j\alpha}(i\xi) = [a(i\xi)]/[3\pi^{3/2} a^3].$$

The traditional way to treat these integrands is to make a series expansion of the logarithm in Eq. (4) and keep only the lowest-order term, i.e., $\ln(1 + x) \approx x$. However, the energy from the eigenmodes for an isotropic atom in vacuum is given by the sum of the equal $j = x, y,$ and $z$ contributions from the


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TABLE I. Finite-size dispersion self-energy, $E_s$, for noble gas atoms and two ions. The subscript expanded indicates that the result is from using series expansion of the logarithm in the integrand. All energies are in eV. The input data were taken from Refs. [5–8].

<table>
<thead>
<tr>
<th>Element</th>
<th>$E_s^{\text{NR, full}}$</th>
<th>$E_s^{\text{NR, expanded}}$</th>
</tr>
</thead>
<tbody>
<tr>
<td>He</td>
<td>71.2</td>
<td>131.5</td>
</tr>
<tr>
<td>Ne</td>
<td>104.9</td>
<td>220.3</td>
</tr>
<tr>
<td>Ar</td>
<td>37.6</td>
<td>62.1</td>
</tr>
<tr>
<td>Kr</td>
<td>29.9</td>
<td>47.5</td>
</tr>
<tr>
<td>Na$^+$</td>
<td>20.2</td>
<td>22.4</td>
</tr>
<tr>
<td>Cl$^-$</td>
<td>5.2</td>
<td>6.0</td>
</tr>
</tbody>
</table>

The difference between perturbative and nonperturbative theories will be much reduced in a dielectric medium due to the factor $1/\epsilon$ (where $\epsilon$ is the frequency-dependent dielectric function of a medium). Hence the results obtained for instance for the permeability of atoms across a membrane will be changed mainly due to changes in the self-energy in vacuum. A factor of 2 difference for these self-energies compared to those obtained from a series-expanded theory ought therefore to be measurable with existing experimental equipment for solvation free energies and permeabilities. Large corrections have been overlooked in the past when performing series expansions of the logarithmic terms before including the finite size. The effects of a nonperturbative theory ought in principle to be detectable experimentally, although it is a challenge to measure self-energies directly. What is practically possible is to verify our results indirectly in experiment. Solvation energies of atoms and ions in a dielectric medium (i.e., changes of the self-energy in a vacuum compared to in a medium) can be measured [9–11]. Latimer et al. [12] were able to fit experimental solvation energies (or rather the related heats of solvation) to the Born equation by increasing the effective radius of the ions. Self-energy changes have also been shown to influence permeabilities across dielectric membranes [13,14].

We give in Table I the results found when using our nonperturbative theory and as comparison also the results from using the approximate series-expanded theory. As input we have used atomic radii and static polarizabilities given by Hohm and Thakkar [5] and characteristic frequencies given by Mahan and Subbaswamy [6]. For the ions we use the input data from Parsons and Ninham [7,8].

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